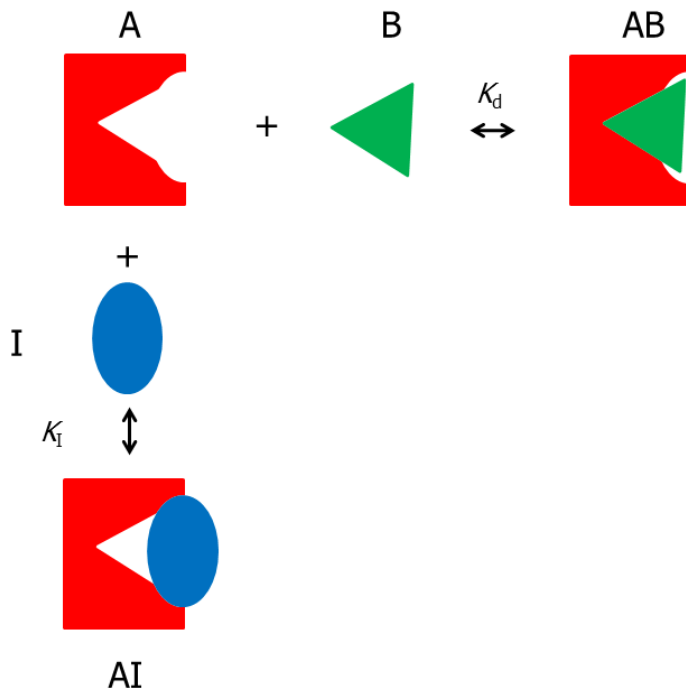


## Competitive inhibition model



Reactant B binding to reactant A is competitively inhibited by inhibitor I

$K_d$  = Dissociation constant of A and B binding

$K_i$  = Dissociation constant of A and I binding

In the reaction sample you know...

Total reactant A concentration =  $A_0$

Total reactant B concentration =  $B_0$

Total inhibitor I concentration =  $I_0$

However, you do not know...

Free reactant A concentration

Free reactant B concentration

Free inhibitor I concentration

Free product AI concentration

Free product AB concentration

Suppose you can indirectly determine concentration of product AB by a method which gives you values (activity, optical density, peak height, etc.) proportional to the concentration.

(Activity value) =  $k_1 \times [AB]$  (concentration).

If AI complex also contribute activity values (usually not!), actual activity values are

$$(\text{Activity value}) = k_1 \times [\text{AB}] (\text{concentration}) + k_2 \times [\text{AI}] (\text{concentration})$$

By performing curve fitting you try to obtain...

$K_d$ ,  $K_i$ ,  $k_1$ , and  $k_2$  values

In this example (data sets) concentration of  $A_0$  is fixed. Each data set was generated after measuring activity values in the presence of fixed concentration of  $B_0$  after adding various concentration of  $I_0$ .

independent variable (x1 values) is  $I_0$

dependent variable (y values) is activity.

There are two equations in each data set (total of four sets).

$$y = \frac{k_1 \cdot B_0 \cdot A}{(K_d + A)} + \frac{k_2 \cdot I_0 \cdot A}{(K_i + A)}$$

$$A_0 - A - \frac{B_0 \cdot A}{(K_d + A)} - \frac{I_0 \cdot A}{(K_i + A)} = 0$$

(Newton Raphson equation)

Press "**Fill Sheet with selected example data**" button to generate a new sheet filled with example data sets.

Described below are the input texts of equations according to "Curve Fitter Excel Add-in" syntax.

DataSet 1 ( $A_0 = 20$ ,  $B_0 = 20$ )

$$y = p_3 \cdot 20 \cdot n_1 / (p_1 + n_1) + p_4 \cdot x_1 \cdot n_1 / (p_2 + n_1)$$

$$f = 20 - n_1 - 20 \cdot n_1 / (p_1 + n_1) - x_1 \cdot n_1 / (p_2 + n_1)$$

DataSet 2 ( $A_0 = 20$ ,  $B_0 = 40$ )

$$y = p_3 \cdot 40 \cdot n_1 / (p_1 + n_1) + p_4 \cdot x_1 \cdot n_1 / (p_2 + n_1)$$

$$f = 20 - n_1 - 40 \cdot n_1 / (p_1 + n_1) - x_1 \cdot n_1 / (p_2 + n_1)$$

DataSet 3 ( $A_0 = 20$ ,  $B_0 = 60$ )

$$y = p_3 \cdot 60 \cdot n_1 / (p_1 + n_1) + p_4 \cdot x_1 \cdot n_1 / (p_2 + n_1)$$

$$f = 20 - n_1 - 60 \cdot n_1 / (p_1 + n_1) - x_1 \cdot n_1 / (p_2 + n_1)$$

DataSet 4 ( $A_0 = 20$ ,  $B_0 = 80$ )

$$y = p_3 \cdot 80 \cdot n_1 / (p_1 + n_1) + p_4 \cdot x_1 \cdot n_1 / (p_2 + n_1)$$

$$f = 20 - n_1 - 80 \cdot n_1 / (p_1 + n_1) - x_1 \cdot n_1 / (p_2 + n_1)$$

$$K_d = p_1$$

$$K_i = p_2$$

$$k_1 = p_3$$

$$k_2 = p_4$$

free reactant A concentration =  $n_1$  (determined by Newton Raphson method)

\*In reality determine  $p_1$  and  $p_3$  values by fitting data set in the absence of inhibitor (I) separately. (Also determine  $p_4$  by an experiment in the absence of reactant B.) Then plug-in those value to the above equation and obtain  $p_2$  value by curve fitting.